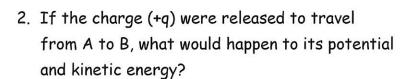
Notes - 19.1 Electric Potential Energy: Potential Difference

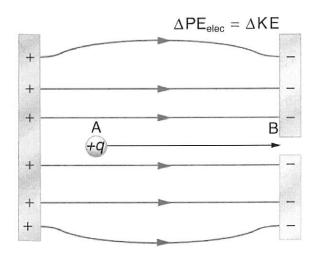
The diagram on the right presents an analogy comparing electric potential energy to gravitational potential energy. Gravitational and Electrostatic forces are both conservative, so the law of conservation of energy applies.

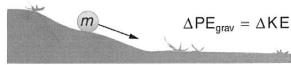
1. Where in the diagram would the positive charge have the greatest potential energy?

Left; near positive plate



PE: <u>Pecrease</u> KE: <u>Increase</u>





Total Energy conserved

- 3. *Electric Potential* is electric potential energy per unit of charge. This is not the same as electric potential energy. It is a ratio that applies to a specific point in an electric field, and it allows us to determine the potential energy of any charge at that point.
- 4. Voltage = Electrical Potential Difference. This is the difference between the electrical potentials at two different points.

5. Units and symbols:

Quantity	Symbol	Formula	Units
Electrical Potential	(PE 11	aV.	National Control of Co
Energy	-clea)	7.	
Electrical Potential	V	PE 2	7
Voltage (Electrical Potential	VAV	A PE	Section 2011
Difference)		Z	C

0.	in practice,	you can	use tr	iese t	•
	11			1	

6. In practice, you can use these formulas and units:

$$\Delta PE_e = 2V \qquad units \Rightarrow J \qquad use this$$

$$V \text{ (voltage)} = \Delta PE \qquad units \Rightarrow J \qquad vsethis$$

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7. Example Problem:

Suppose you have a 12.0 V motorcycle battery that can move 5000 C of charge, and a 12.0 V car battery that can move 60,000 C of charge. How much energy does each deliver? (Assume that the numerical value of each charge is accurate to three significant figures.)

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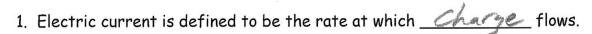
$$PE_{e-mobrayde} = 9V = 5000c(12V) = 60,000 J$$
 $PE_{e-mobrayde} = 60,000c(12V) = 72,000 J$

8. Electron Volts: It is useful to have an energy unit related to submicroscopic effects. An energy unit called the electron volt (eV), which is the energy given to a fundamental charge accelerated through a potential difference of $\frac{1}{\sqrt{2}}$

$$1eV = \frac{1.6 \times 10^{-19}}{J}$$

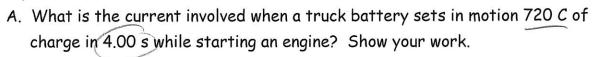
$$PE = (1.6 \times 10^{-19})(1V) = 1$$

Notes - 20.1 Current





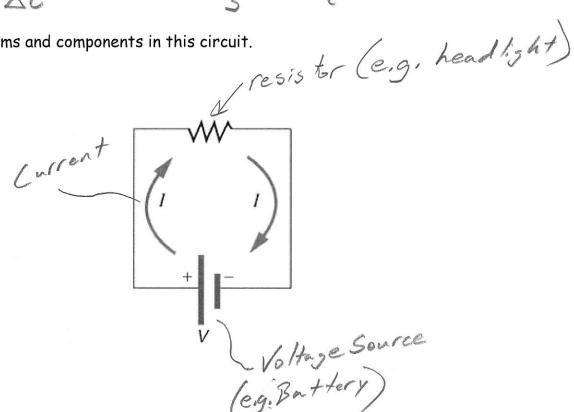
Example:

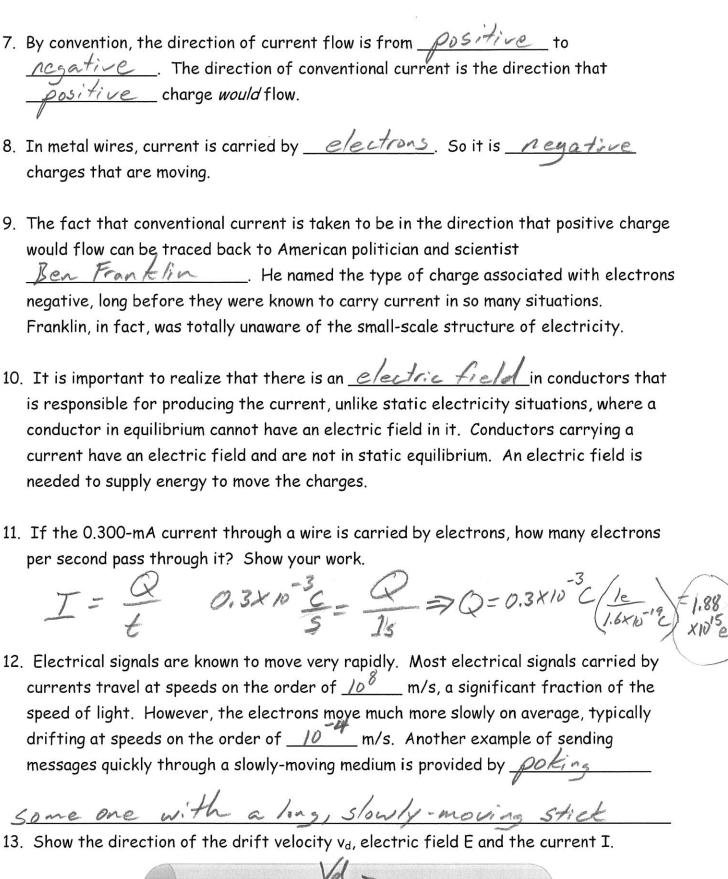


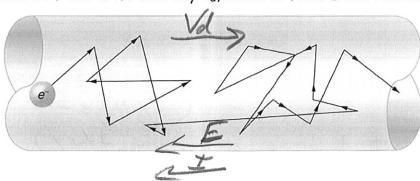
B. How long does it take 1.00 C of charge to flow through a handheld calculator if a 0.300-mA current is flowing? Show your work.

$$I = \frac{\Delta Q}{Xt} = \frac{0.3 \times 10^{-3} \text{ c}}{5} = \frac{10}{t} = \frac{10}{5} = \frac{3.33 \times 10^{3} \text{ s}}{5}$$

6. Label the terms and components in this circuit.







Practice - 19.1 Electric Potential Energy: Potential Difference

 $q_{electron} = -1.6 \times 10^{-19} C$ $KE = \frac{1}{2} \text{ mv}^2$

1. A. What is the speed of an electron starting from rest accelerated through a

potential difference of 100 V? $m_e = 9.11 \times 10^{-31} \text{ kg}$ KE = APE = QV = 1.6×10-19 (100 I) = 1.6×10-13

KEFinal = /2mv2 1.6×10 9=1/2 (9.11×103/2) V2

B. What is the speed of a proton starting from rest accelerated through a potential difference of 100 V? $m_p = 1.67 \times 10^{-27} \text{ kg}$ Same final KE aspart A -> 1.6x 10-175

KE= 1/2mv2 => 1.6x10 15=1/2 (1.67x10 1/2) V2 (V=1.38×10 5m/k

2. An evacuated tube uses an accelerating voltage of 40.0 kV to accelerate electrons to hit a copper plate and produce x rays. What is the speed of these electrons?

KEFIND = OPE = 9V = (1.6x10-12) 40×10-1 = 64×10-165 KE=1/2 => 64x10 15 = 1/2 (9,11x10 3/2) V

- 3. A bare helium nucleus has two positive charges and a mass of 6.64×10^{-27} kg.
 - A. Calculate its kinetic energy in joules at 2.00% of the speed of light. $c = 3.00 \times$ KE = /2 (6.64× 10 1/3) (0.02) 3×108 m/s)

(= 1,2×10-13)

B. What is this in electron volts?

1,2×10-13- (1ex 195) = 7.5×10°eV

C. What voltage would be needed to obtain this energy?

1.2×10-135 = (2)(1.6×10-19) V =7(V=3,75×10 5V SPE=2V

1. Car batteries are rated in ampere-hours ($A \cdot h$). To what physical quantity do ampere-hours correspond (voltage? Charge? Energy? . . .)?

Ah = Charge (time) = Charge

5. What is the current in milliamperes produced by the solar cells of a pocket calculator through which 4.00 C of charge passes in 4.00 h?

I = 40 = 40 = 2.78×10 A

6. What is the current when a typical static charge of 0.250 μ C moves from your finger to a metal doorknob in 1.00 μ s?

 $T = \frac{\Delta Q}{\Delta t} = \frac{0.25 \times 10^{\circ} C}{1 \times 10^{\circ} S} = \frac{0.25 A}{1 \times 10^{\circ} S}$

7. A large lightning bolt had a 20,000-A current and moved 30.0 C of charge. What was its duration?

 $I = \begin{cases} 20,000A = \frac{30C}{t} & t = 1.5 \times 10^{\frac{3}{5}} \end{cases}$

8. A defibrillator passes 12.0 A of current through the torso of a person for 0.0100 s. How much charge moves?

 $I = \frac{Q}{t} \quad 12A = \frac{Q}{0.015} \Rightarrow Q = 0.12C$

- 9. A clock battery wears out after moving 10,000 C of charge through the clock at a rate of 0.500 mA.
 - A. How long did the clock run?

 $I = \frac{Q}{t} \quad 0.5 \times 10^{-3} A = \frac{19000C}{t} \Rightarrow t = Z \times 10^{-5} S$

B. How many electrons per second flowed?

 $I = 0.5 \times 10^{-3} \text{ C}$ $\left(\frac{1e^{-19}}{5}\right)^{-19} = 3.13 \times 10^{-15} \text{ C}$

Solutions:

charge

5. 0.278 mA

9. A. 2.00×10^7 s B. 3.13×10^{15} e⁻/s

6. 0.250 A

7. 1.50×10^{-3} s

8. 0.120 C